

Conformation and Ring Topomerization in *cis-syn-cis*-2,2,6,6-Tetramethoxyperhydroanthracene¹

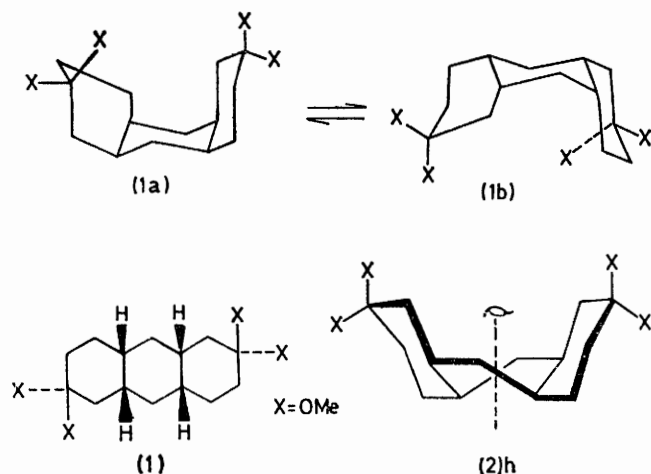
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Summary The free activation energy for ring topomerization in *cis-syn-cis*-2,2,6,6-tetramethoxyperhydroanthracene at 212 K equals 47 kJ mol⁻¹. RING inversion in cyclohexane derivatives has been extensively studied. Results for *cis*-decalin systems are scarce, and to our knowledge there are no reports on inverto-

mers of perhydroanthracenes, possibly because they are difficult to prepare, although they are interesting because three rings are involved in the topomerization² process.

We have studied the temperature dependence of the ¹H n.m.r. spectrum of *cis-syn-cis*-2,2,6,6-tetramethoxyperhydroanthracene (**1**) (Varian HA-100, 10% in CS₂), prepared from the known³ *cis-syn-cis*-perhydroanthracene-2,6-dione (m.p. 130 °C). At room temperature the methoxy-signals consist of two sharp, equally intense peaks. At lower temperatures ($T_{\text{coal.}} = 212$ K) there are four equally intense, but partially overlapping methoxy-signals. The splitting $\delta_{\text{max.}} = 5.6$ Hz for one of the singlets to a doublet on lowering the temperature could easily be followed and



leads to $\Delta G_{212}^\ddagger = 47$ kJ mol⁻¹. Methoxy-signals have very favourable relaxation times and, despite the relatively small low-temperature shift difference, the condition $\Delta\nu \gg 1/T_2$ is met.

We favour the view that an all-chair/all-chair interconversion occurs, *e.g.* (**1a**) \rightleftharpoons (**1b**). In this conformation, compound (**1**) has no symmetry element, so the four methoxy-groups are diastereotopic⁴ and are found to be anisochronous under slow exchange conditions. Under fast exchange conditions (hypothetically planar rings) the effective point group of (**1**) is C₂ (axis perpendicular to the plane of the paper). There is a diastereotopic pair of equivalent methoxy-groups, and two signals are indeed found.

On the other hand the all-chair form of *cis-syn-cis*-perhydroanthracene has been calculated⁵ to be only 8.6 kJ mol⁻¹ more stable than a conformation in which the central ring assumes a classical boat form (**2**). Compound (**1**), in its all-chair form, moreover is strained by a 1,3-diaxial MeO \cdots CH₂ interaction, and this strain has been estimated⁶ as high as 8 kJ mol⁻¹. This interaction can be lowered by having the central ring in a boat form (non-classical boats are less likely here, because the torsion angles of a classical boat only fit an annellation with chair forms). A good case can be made for accepting that the central ring in (**1**) exists in the boat form, but in such a conformation (**2**) has a C₂ axis, and cannot give rise to the four different methoxy-signals in the n.m.r. spectrum.

Finally, the barrier to inversion in *cis*-2,2,7,7-tetramethoxydecalin is higher ($\Delta G_{238}^\ddagger = 52$ kJ mol⁻¹)⁷ than in (**1**), despite the fact that in the latter compound three rings are involved in the topomerization, and only two in the former. This can be ascribed to the severe 1,3-diaxial strain in the all-chair form of (**1**) which raises the energy of its ground state.

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¹ For previous papers in the series: 'NMR-Experiments on Acetals' see *Bull. Soc. chim. belges*, 1973, **82**, 413 and *Tetrahedron Letters*, 1973, 75.

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